

Synthesis and Characterization of 1,2-Dihydroquinoline Sulphonamides: Novel Derivatives with Enhanced Pharmacological Potential

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Abstract

The present investigation reports the synthesis and characterization of a novel series of 1,2-dihydroquinoline sulphonamides, developed with the aim of enhancing their pharmacological efficacy. The synthesis involved a multi-step process starting with the preparation of 3-hydroxy Acetanilide, followed by Pechmann condensation to obtain acetylated aminophenol, and subsequent reactions leading to the formation of 1,2-dihydroquinoline derivatives. These derivatives were further reacted with various substituted sulphonyl chlorides to yield the target sulphonamide compounds. The synthesized compounds were characterized using FT-IR, ¹H NMR, ¹³C NMR, and LCMS. Their pharmacological activities, including antibacterial, anthelmintic, anti-inflammatory, and antihyperlipidemic properties, were evaluated through in-vitro assays.

The results demonstrated that these novel 1,2-dihydroquinoline sulphonamides exhibit promising pharmacological activities, validating their potential as effective therapeutic agents.

Keywords: 1,2-Dihydroquinoline, Sulphonamides, Synthesis, Characterization, Pharmacological Activity, Antibacterial, Anthelmintic, Anti-inflammatory, Antihyperlipidemic

Introduction

Quinoline is the most common heterocyclic compound in medicinal chemistry. Quinoline chemistry has been increasing interest as most of these substances are useful as chemotherapeutic agents for malaria and microbes [1, 2]. Quinoline was first isolated by Ferdinand Rungin 1834 from the coal tar bases (very high viscosity brown or black liquid) [3]. Gerhardt later in 1842 obtained quinoline by alkaline pyrolysis of cinchonine, alkaloid analogue of quinoline. It was found that heterocyclic compounds containing nitrogen and oxygen are one of the most commonly synthesized and evaluated compounds as they possess pharmacologic activities including antihyperlipidemic and antimicrobial activity.

Quinolines are an important class of natural flavonoids [4]. These are biological agents,

including antibiotic [5], anthelmintic [6], anti-inflammatory [7], anti-fungal [8-9], anti-tumour [10-11] and antimitogenic [12]. In addition, some quinoline derivatives have inhibited many enzymes such as xanthine oxidase [13], protein tyrosine kinase [14, 15] in cell systems. Therefore, the synthesis of quinolines is of great interest.

Quinolines are a group of synthetic wide spectrum antibacterial drugs. Nalidixic acid was a first generation quinoline introduced in 1962 for the cure of urinary tract infections in humans.

George Leisher and coworkers discovered Nalidixic acid in a distillate during an attempt at synthesis of famous antimalarial drug chloroquine.

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Quinoline derivatives exhibited their antibacterial effect by preventing bacterial DNA replication in bacterial cell, however most of the quinolones in clinical use belong to the fluoroquinolones, which have fluorine atom attached to benzene ring. Therefore, we decided to synthesize quinoline derivatives and evaluate them for their activities for instance Antibacterial, Anthelmintic, Anti-inflammatory and Anti hyperlipidemic activity. In literature, reviews on importance of alkylation reactions [16-18], formulations [19,20] and dosage [21] are well documented. This gave an impetus to the development of newer pharmacologically active quinoline derivatives. Hence, we synthesized some of the novel quinoline derivatives as shown in the reaction scheme 2.1 and to screen them for their activities. Sanchez JP *et al.* [22] prepared a new series of amino sulfonamide and benzene sulfonamide containing substituted piperazine quinoline carboxylic acids with (1, 2) at C-3 position and screened against various bacterial strains, using cup plate method. Compounds revealed promising antibacterial activity against tested bacterial strains. Ghorab *et al.* [23] described the preparation of few new derivatives of quinoline having quinoline hydrazone motif. These hydrazones act as CA inhibitors, which may be the reason for their Anti hyperlipidemic activity. In vitro anti-cancer activity against MCF7 (breast cancer cell line) results show that compound (3) having tetrahydroquinoline moiety was more potent. Lunniss *et al.* [24] developed selective PDE4 inhibitor quinolines (4a, 4b) with utility in chronic Smith and coworkers [25] developed Quinoline based NK3 receptor antagonists (5) and (6) with CNS activity. Verma *et al.* [26] prepared new derivatives of quinoline-sulfonamide hybrids (7) as antimalarial. The results showed that the derivatives exhibited good activity. Jian *et al.* [27] described the synthesis of a novel series of quinoline derivatives as potent c-Jun N-terminal kinase (JNK) inhibitors (8) with selectivity against p38. The results revealed that compound possessed a remarkable inhibition activity ($IC_{50}=0.15M$). Chloroquinolone derivatives have been synthesized by Kovi *et al.* [28] the

compound (9) exhibited potent antimalarial activity at submicro molar concentration levels. Hu *et al.* [29] prepared pyridinylquinoline derivatives with PI3K α inhibition activity. Amongst all, compound (10) showed exceptional inhibition of PI3K α . In another research, the same group have reported unsaturated-sulfonylamino-quinolines (11) as PI3K α inhibitors. All the compounds reported by them were tested for PI3K α inhibitory activity and the results showed that they possess better activity than BEZ235, the positive control. Chen *et al.* [30] have filed a patent on a class of morpholino-quinoline sulphonamides useful in treating the disease associated with PI3K/mTOR cancer. One of the representative compounds (12) effectively inhibited with $IC_{50} < 20nM$. Brown *et al.* [31] patented novel series of amino substituted quinoline sulphonamides. All the compounds were tested for lactate dehydrogenase A inhibitory activity and they reported that the compound (13) exhibited a pIC_{50} value of 7.1 in no less than one set of experimental runs. Lee *et al.* [32] prepared quinoline sulphonamide derivatives as Anti hyperlipidemic agents. Compound (14) presented good Anti hyperlipidemic activity against the MDA-MB231, MDAMB468 and MCF-breast cancer cell lines. Mohsen Nikpour [33] research group designed and synthesized methyl benzamide derivatives and evaluated them for their inhibitory activity against human PDE3A and PDE3B. The best activity was obtained for the compound (15) ($IC_{50} = 0.43 \pm 0.041M$). obstructive pulmonary disorder. Modather F *et al.* [34] planned and synthesized and characterized novel series of methylquinolin(1H)-one derivatives. In order to find good antioxidant agents, efficiency in antioxidant activity of the synthesized compounds have been investigated using ASTMd-942 and ASTMd-664. Good activity was observed in the compound (16). P. Raghavendra *et al.* [35] synthesized a bunch of substituted benzylidene-1,2-dihydroquinoline compounds by condensing substituted imidazole with substituted quinoline. The

compounds were inspected for anti-inflammatory and its ulcerogenicity activities. They reported that all the lead compounds (**17**) showed better activities against inflammation when compared with Ibuprofen.

Jumbad Hetal. [36] synthesized a series of new quinoline-2-one Schiff bases containing isoxazolin nucleus. Some of the synthesized compounds have been screened for antibacterial activities against two types of bacteria *E. coli* and *S. aureus*. Compound (**18**) showed good activity against *S. aureus* but not against *E. coli*.

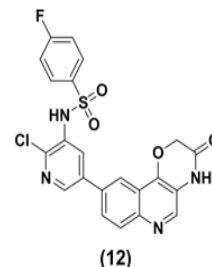
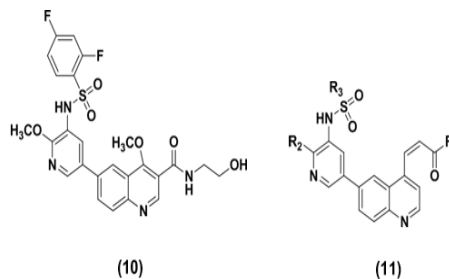
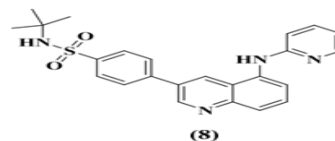
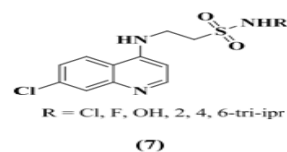
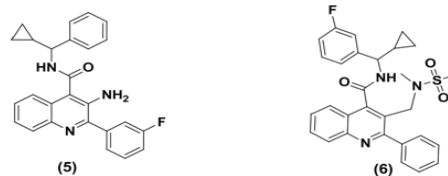
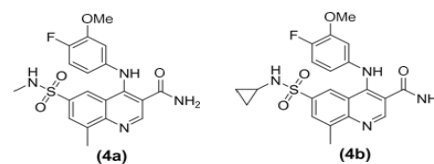
MBDeshmukand coworkers [37] synthesized the compound 7-hydroxy-4-methylquinolin-2(1H)-one (**19**) by an easy and efficient microwave-assisted method. Compound was successfully characterized on the basis of its ^1H proton and ^{13}C carbon magnetic measurements and elemental analysis. The compound was also subjected for antibacterial and antifungal study.

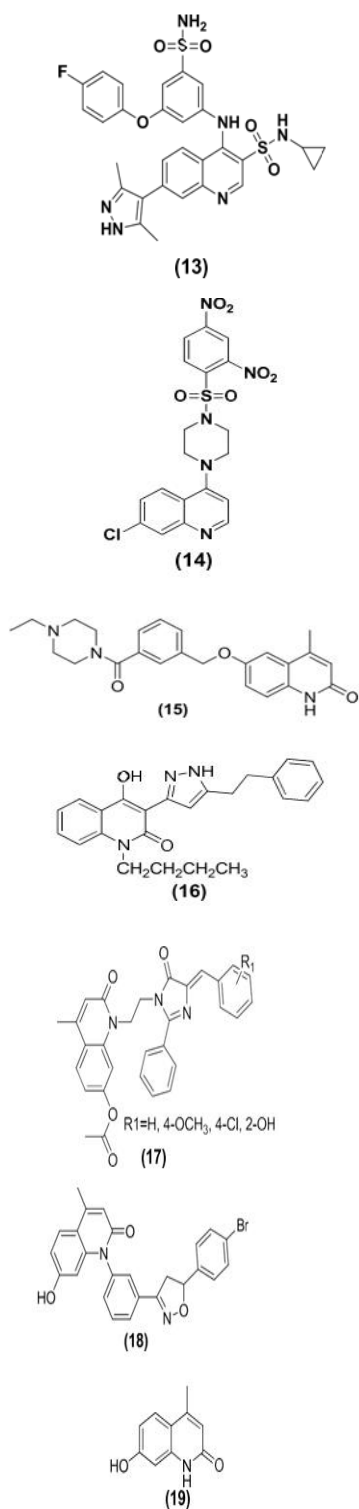
Based on the survey of literature on quinolone containing sulphonamides and compounds containing 1,2-dihydroquinolinone nucleus, it was found that compounds

are pharmacologically very important therefore the objective of present work was focused to plan and synthesize 1,2-dihydroquinolone derivatives containing sulfonamide moiety. Finally, all the synthesized derivatives were well characterized and evaluated them for their *in-vitro* Antimicrobial, Anthelmintic, Anti-inflammatory and Antihyperlipidemic study. *In-silico* study was also performed to infer the binding pattern of synthesized compounds.

The objective of the present synthetic strategy was to synthesis 1,2-dihydroquinolone sulphonamide derivatives [**4a-h**]. The synthetic route for the preparation is based on the transformation of 3-hydroxy Acetanilide into acetylated aminophenol followed by cyclized product.

The synthesis and characterization of 1,2-dihydroquinolone sulphonamides [**4a-h**] follows the below mentioned sequence of reactions:



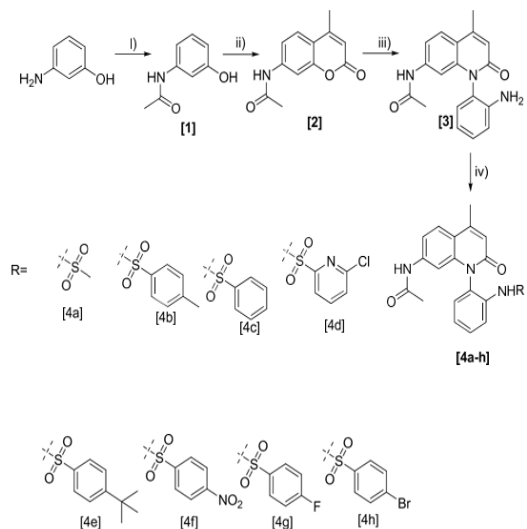


Step-1: Synthesis of 3-hydroxy Acetanilide [1] was carried out by the reaction of 3-aminophenol in presence of acetic acid with acetic anhydride.

Step-2: 3-hydroxy Acetanilide [1] underwent Pechmann condensation reaction with ethylacetoacetate to afford product acetylated aminophenol [2].

Step-3: Reaction of O-phenylenediamine with intermediate acetamide rearrangement takes place leading to ring opening and subsequent elimination of water afforded condensed product [3].

Step-4: Compound [3] on reaction with substituted Sulphonamides in presence of base underwent coupling reaction to obtain sulphonamides containing dihydroquinoline derivatives [4a-h]. This sequence of reaction for the synthesis of 1,2-dihydroquinoline sulphonamides [4a-h] has been presented in scheme-2.1.



Scheme-2.1: Synthetic reaction scheme for the preparation of 1,2-Dihydroquinoline Sulphonamides [4a-h]

Material and Methods

Materials

Organic solvents and Reagents

The organic solvents were purchased from SD Fine, Spectrochem, Sigma Aldrich and standard commercial sources are used without further purification.

Reagents and conditions:

i) $(\text{CH}_3\text{CO})_2\text{O}$, CH_3COOH , 8h; ii) $\text{CH}_3\text{COCH}_2\text{COOEt}$, 70% H_2SO_4 , 0°C , 9-10h; iii) *O*-phenylenediamine, NaOAc, rt, 8h; iv) Substituted Sulfonyl chlorides, Et_3N , CH_2Cl_2 .

Reaction conditions

Room temperature reactions mentioned ranges between $20\text{-}30^\circ\text{C}$ (throughout the year). For low temperature reactions, ice-bath with sodium chloride (-5°C) was used. Magnetically stirred oil bath (or) a hotplate was used for high temperature reactions.

Analytical techniques

Melting point

For solid compounds melting point range was determined in open capillary tubes using a digital melting point apparatus.

Thin layer Chromatography (TLC)

Reaction was monitored by TLC using silica plates. The following mobile phases were used for elution: Hexane: Ethyl acetate (or) dichloromethane: methanol in different ratios. TLC plates were visualized with ultraviolet light or iodine vapors or by staining with 2% aqueous potassium permanganate solution.

Column chromatography

Silica gel (60-120 mesh) or neutral alumina was used for purification of synthesized compounds.

Instrumentation

Compounds were characterized by available spectroscopic techniques and LCMS.

FT-IR

The spectra were recorded using JASCO FTIR-4100 spectrophotometer as KBr disc in the range of $4000\text{-}400\text{ cm}^{-1}$. FT-IR data are reported as follows: Frequency (functional group).

^1H NMR and ^{13}C NMR

The ^1H and ^{13}C NMR spectra were recorded on JEOL-400 MHz spectrometers using deuterated solvents (CDCl_3 & $\text{DMSO-}d_6$) and

Tetramethylsilane (TMS) as an internal standard. The chemical shift values were expressed in δ ppm in both ^1H NMR (0-15 ppm) and ^{13}C NMR (0-200 ppm) spectra.

^1H NMR data is reported in the following order: chemical shift (multiplicity, *J* value, number of protons and nature of proton). ^{13}C NMR data are reported in the following order: chemical shift (numbered carbon atom).

Mass analyses

Waters Alliance 2795 separation module and Waters Micromass LCT mass detector was used to record LC-MS.

Elemental analysis

Elemental analysis (C, H and N) was performed on Elementar vario MICRO cube.

Experimental

Step-1: Synthesis of 3-hydroxyacetanilide [1]

3-aminophenol (25g, 0.11 mol) taken in round bottom flask containing acetic anhydride (80 mL) and it was stirred for 8 h at 60°C under Nitrogen atmosphere. After reaction completion, excess acetic anhydride was removed under reduced pressure; the residue was obtained which was dissolved in dichloromethane. The dichloromethane layer was washed with brine solution, dried over anhydrous Na_2SO_4 and concentrated to achieve compound [1]. LCMS: 152 (M+1), MP: 152°C . Yield: 82%.

Step-2: Synthesis of intermediate acetamide [2]

3-hydroxyacetanilide [1] (25g, 0.11 mol) was taken in ethyl acetate (0.1 mol) with 70% aq. H_2SO_4 (50 mL), and it was stirred for 9-10 h at 0°C . After reaction completion, the reaction mixture was poured into ice cold water, solid separates out. The solid was filtered and washed with water. The crude product was recrystallized to obtain pure compound [2].

Step-3: Synthesis of N-[1-(2-aminophenyl)-4-methyl-2-oxo-1,2-dihydroquinolin-7-yl]acetamide [3]

Compound [2] (2.17g, 0.01 mole), *o*-phenylenediamine (0.01 mole, 1.08g) and sodium acetate (5 g) were taken in a round bottom flask containing glacial acetic acid (15 mL) and the resulting mixture was refluxed for 8 h. After completion reaction,

thereactionmixturewasbringdowntoroomtemperat ure. These separated solid was filtered and recrystallized from methanol:water (1:2) to give title compound [3]. LCMS: 237.8 (M+1), MP: 285°C. Yield: 70%.

Step-4: Synthesis of 1,2-

dihydroquinoline Sulphonamides [4a-h]

Equimolar quantities of compound [3] (0.5g, 0.001 mol), different substituted sulphonyl chlorides (0.001 mol) and triethylamine (0.57g, 0.003 moles) were stirred in dry methylene chloride (10 mL) under N₂ atmosphere at room temperature for 12 h. The reaction was monitored by TLC. After reaction completion, the mixture was washed with water and brines solution. Methylene chloride layer was dried over anhydrous Na₂SO₄ and evaporated under vacuum. The residue was purified by column chromatography using petroleum ether: ethyl acetate as eluent (7:3) to get Sulphonamide dihydroquinoline nucleus [4a-h] in good yield (scheme 2.1).

Results and Discussion

In this work, pharmacologically important novel dihydroquinoline derivatives were synthesized by the acetylation reaction of 3-Aminophenol, cyclisation of acylated aminophenol acetamide followed by substitution of *o*-phenylenediamine to get intermediate [3]. Compound [3] was reacted with various substituted sulphonyl chlorides in order to get the dihydroquinoline derivatives with sulphonamide moiety [4a-h].

The structures of the newly synthesized compounds were established by spectroscopic methods for instance FT-IR, ¹H-NMR, ¹³C-NMR and LCMS or Mass spectra.

Finally, all the synthesized compounds were evaluated for their *in-vitro* Antibacterial, Anthelmintic, Anti-inflammatory and Antihyperlipidemic activity.

Spectral interpretation of the title compounds (4a-h)

N-[1-(2-Methanesulfonylamino-phenyl)-4-methyl-2-oxo-1,2-dihydro-quinolin-7-yl]acetamide [4a]

IR: $\nu_{\text{max}}/\text{cm}^{-1}$: 3340 (N-H), 2228 (CN), 1698 (CO), 1342–1140 (CF stretching); **¹H-NMR (CDCl₃)** δ : 7.34–7.28 (d, *J*=8.7 Hz, 1H, Ar-H), 7.26–7.12 (m, 7H, Ar-H & NH), 6.98–6.96 (d, *J*=8.3 Hz, 1H, Ar-H), 6.73–6.71 (d, *J*=8.3 Hz, 1H, Ar-H),

2.98 (s, 3H, COCH₃), 2.36 (s, 3H, CH₃), 2.16 (s, 3H, CH₃); **¹³C-NMR**

(CDCl₃) δ : 163.25, 135.75, 132.29, 129.97, 129.40, 12.92, 118.34, 114.30,

55.57, 28.27, 22.27; **MS:** *m/z* = Cal. 385.44 (Found 386.1) (M+1); **CHN**

analysis: Calculated for: C₁₉H₁₉N₃O₄S; Calculated: C, 59.21%; H, 4.97%; N, 10.90%; Observed: C, 59.19%; H, 4.95%; N, 10.88%.

N-[4-Methyl-2-oxo-1-[2-toluene-4-sulfonylamino)-phenyl]-4-methyl-2-oxo-1,2-dihydro-quinolin-7-yl]-acetamide [4b]

IR: $\nu_{\text{max}}/\text{cm}^{-1}$: 3340 (N-H), 2228 (CN), 1698

(CO), 1342–1140 (CF stretching); **¹H-NMR (CDCl₃)** δ : 7.74–7.70 (m, 4H, Ar-H), 7.35–7.31 (m, 4H, Ar-H), 7.27–

7.26 (d, *J*=8.3 Hz, 1H, Ar-H), 6.99–6.97 (t, 2H, Ar-H), 6.96–6.88 (m, 5H, Ar-H), 2.58

(s, 3H, COCH₃), 2.36 (s, 3H, CH₃), 2.16 (s, 3H, CH₃); **¹³C-NMR (CDCl₃)**

δ : 173.28, 170.45, 163.80, 161.33, 154.55, 130.71, 130.63, 126.32, 126.24, 123.55, 123.52, 119.57, 119.35, 114.90, 114.66, 107.79, 80.08, 43.10, 28.27, 22.27;

MS: *m/z* = Cal. 461.53; Found 462.2 (M+1);

Elemental analysis: Calculated for: C₂₅H₂₃N₃O₄S; Calculated: C, 65.06%; H, 5.02%; N, 9.10%; Observed: C, 65.04%; H, 5.00%; N, 9.08%.

N-[1-(2-benzenesulfonylamino-phenyl)-4-methyl-2-oxo-1,2-dihydro-quinolin-7-yl]-acetamide [4c]

IR: $\nu_{\text{max}}/\text{cm}^{-1}$: 3340 (N-H), 2228 (CN), 1698 (CO), 1342–1140 (CF stretching); **¹H-NMR (CDCl₃)** δ : 7.43–7.29 (d, *J*=8.7 Hz, 4H, Ar-H), 7.22–7.18 (m, 8H,

Ar-H & NH), 6.96–6.94 (d, *J*=8.3 Hz, 1H, Ar-H), 6.72–6.69 (d, *J*=8.3 Hz, 1H, Ar-H),

2.53 (s, 3H, CH₃), 2.17 (s, 3H, CH₃); **¹³C-NMR (CDCl₃)** δ :

169.3, 141.2, 138.3, 134.8, 128.7 CF₃, 125.9, 124.6, 115.9, 104.9, 28.7, 27.8; HPLC: 100% purity; **MS:** *m/z* = Cal. 447.51; Found 448.0 (M+1); **Elemental analysis:** Calculated for: C₂₄H₂₁N₃O₄S; Calculated: C, 64.41%; H, 4.73%; N, 9.39%; Observed: C, 64.39%; H, 4.71%; N, 9.37%.

N-{1-[2-(6-Chloro-pyridine-2-sulfonylamino)-phenyl]-4-methyl-2-oxo-1,2-dihydro quinolin-7-yl}-acetamide[4d]

IR: $\nu_{\max}/\text{cm}^{-1}$: 3340 (N-H), 2228 (CN), 1698 (CO), 1342–1140 (CF stretching); **$^1\text{H-NMR}(\text{CDCl}_3)\delta$:** 7.43-7.29(d, $J=8.7\text{Hz}$, 4H, Ar-H), 7.22-7.18(m, 8H, Ar-H&NH), 6.96-6.94(d, $J=8.3\text{Hz}$, 1H, Ar-H), 6.72-6.69(d, $J=8.3\text{Hz}$, 1H, ArH), 2.53(s, 3H, CH₃), 2.17(s, 3H, CH₃); **$^{13}\text{C-NMR}(\text{CDCl}_3)\delta$:** 169.3, 141.2, 138.3, 134.8, 128.7, 125.9, 124.6, 115.9, 104.9, 28.7, 27.8; **HPLC:** 100% purity; **MS:** m/z = Cal. 482.94; Found 483.0 (M+1);

Elemental analysis: Calculated for: C₂₃H₁₉N₄O₄SCl; Calculated: C, 57.20%; H, 3.97%; N, 11.60%; Observed: C, 57.18%; H, 3.95%; N, 11.58%.

N-{1-[2-(4-tert-butylphenylsulfonamido)-phenyl]-4-methyl-2-oxo-1,2-dihydroquinolin-7-yl}-acetamide[4e]

IR: $\nu_{\max}/\text{cm}^{-1}$: 3340(N-H), 2228(CN), 1698(CO), 1342–1140(CF stretching); **$^1\text{H-NMR}(\text{CDCl}_3)\delta$:** 8.08(s, 1H, Ar-H), 7.78-7.74(m, 4H, Ar-H), 7.56-7.54(d, $J=8.0\text{Hz}$, 1H, Ar-H), 7.35(s, 1H, Ar-H), 7.34(s, 1H, Ar-H), 7.23(s, 1H, Ar-H), 6.94-6.92(t, 1H, Ar-H), 6.79-6.74(t, 1H, Ar-H), 6.61-6.60(d, $J=4.0\text{Hz}$, 1H, Ar-H), 6.35(s, 1H, ArH), 4.0(s, 1H, NH), 2.42(s, 3H, CH₃), 2.04(s, 3H, CH₃), 1.35(s, 9H, (CH₃)₃); **$^{13}\text{C-NMR}(\text{CDCl}_3)\delta$:** 168.9, 158.6, 154.5, 147.5, 138.8, 137.3, 136.6, 131.6, 130.5, 128.0, 127.3, 125.3, 122.7, 120.7, 119.6, 119.0, 117.2, 111.0, 110.7, 34.7, 31.3, 24.0, 19.0.

N-{4-methyl-1-[2-(4-nitrophenylsulfonamido)-phenyl]-2-oxo-1,2-dihydroquinolin-7-yl}-acetamide[4f]

IR: $\nu_{\max}/\text{cm}^{-1}$: 3340(N-H), 2228(CN), 1698(CO), 1342–1140(CF stretching); **$^1\text{H-NMR}(\text{CDCl}_3)\delta$:** 8.39-8.36 (d, $J=8.0\text{Hz}$, 2H, Ar-H), 8.12-8.10(m, 3H, Ar-H), 7.567-7.52(d, $J=8.0\text{Hz}$, 4H, Ar-H), 7.35(s, 1H, Ar-H), 7.34(s, 1H, Ar-H), 7.23(s, 1H, Ar-H), 6.94-6.92(t, 1H, Ar-H), 6.79-6.76(t, 1H, Ar-H), 6.61-6.60(d, $J=4.0\text{Hz}$, 1H, Ar-H), 6.35(s, 1H, Ar-H), 4.0(s, 1H, NH), 2.42(s, 3H, CH₃), 2.04(s, 3H, CH₃); **$^{13}\text{C-NMR}(\text{CDCl}_3)\delta$:** 168.9, 158.6, 151.1, 147.5, 145.8, 138.6, 137.3, 131.6, 130.5,

128.2, 127.5, 124.2, 122.7, 120.7, 119.6, 117.2, 111.0, 110.7, 24.0, 19.0.

N-{1-[2-(4-fluorophenylsulfonamido)-phenyl]-4-methyl-2-oxo-1,2-dihydroquinolin-7-yl}-acetamide[4g]

IR: $\nu_{\max}/\text{cm}^{-1}$: 3340(N-H), 2228(CN), 1698(CO), 1342–1140(CF stretching); **$^1\text{H-NMR}(\text{CDCl}_3)\delta$:** 8.08 (s, 1H, Ar-H), 7.98-7.96(d, $J=8.0\text{Hz}$, 2H, Ar-H), 7.56-7.54(d, $J=8.0\text{Hz}$, 1H, Ar-H), 7.40-7.34 (m, 4H, Ar-H), 7.23 (s, 1H, Ar-H), 6.94-6.92(t, 1H, Ar-H), 6.79-6.76(t, 1H, Ar-H), 6.61-6.60(d, $J=4.0\text{Hz}$, 1H, Ar-H), 6.35(s, 1H, Ar-H), 4.0(s, 1H, NH), 2.42 (s, 3H, CH₃), 2.04(s, 3H, CH₃); **$^{13}\text{C-NMR}(\text{CDCl}_3)\delta$:** 168.9, 166.1, 159.0, 147.5, 138.6, 137.3, 135.3, 130.7, 130.5, 127.3, 122.7, 120.7, 119.6, 119.0, 117.2, 115.8, 111.0, 110.7, 24.0, 19.0.

N-{1-[2-(4-bromophenylsulfonamido)-phenyl]-4-methyl-2-oxo-1,2-dihydroquinolin-7-yl}-acetamide[4h]

IR: $\nu_{\max}/\text{cm}^{-1}$: 3340(N-H), 2228(CN), 1698(CO), 1342–1140(CF stretching); **$^1\text{H-NMR}(\text{CDCl}_3)\delta$:** 8.08(s, 1H, Ar-H), 7.89-7.88(m, 4H, Ar-H), 7.56-7.54(d, $J=8.0\text{Hz}$, 1H, Ar-H), 7.35(s, 1H, Ar-H), 7.34(s, 1H, Ar-H), 7.23(s, 1H, Ar-H), 6.94-6.92(t, 1H, Ar-H), 6.79-6.76(t, 1H, Ar-H), 6.61-6.60(d, $J=4.0\text{Hz}$, 1H, Ar-H), 6.35(s, 1H, ArH), 4.0(s, 1H, NH), 2.42(s, 3H, CH₃), 2.04(s, 3H, CH₃); **$^{13}\text{C-NMR}(\text{CDCl}_3)\delta$:** 168.9, 159.6, 147.5, 138.6, 138.7, 131.9, 131.6, 130.5, 129.5, 127.5, 127.3, 122.7, 120.7, 119.6, 119.0, 117.2, 111.0, 110.7, 24.0, 19.0.

Conclusion

In conclusion, the study successfully synthesized and characterized a series of 1,2-dihydroquinoline sulphonamides through a well-defined synthetic route. The characterization of these compounds confirmed their chemical structures and purity. Pharmacological evaluations revealed that the synthesized sulphonamides exhibit significant antibacterial, anthelmintic, anti-inflammatory, and antihyperlipidemic activities. These findings underscore the potential of 1,2-dihydroquinoline sulphonamides as versatile pharmacological agents. Future research could explore the mechanisms underlying their biological activities and their efficacy in clinical applications,

potentially leading to the development of new therapeutic options in various medical fields.

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